

Communications

1, 3-Dipolar cycloaddition reactions of
N-cyclohexylnitrone

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Received 6 July 1994; accepted 15 September 1994

1, 3-Dipolar cycloaddition reaction of nitrone **1**, derived from 2, 3-dihydro-4*H*-pyran with different dipolarophiles, have been studied. The reactions have been found to be highly regio- and stereo-selective.

Nitrones are versatile synthetic intermediates and excellent spin trapping reagents¹. Nitrones are prepared either by condensation of aldehydes and ketones with hydroxylamines¹ or by oxidation of the corresponding N, N-disubstituted hydroxylamines².

In the present study the formation of nitrone **1** has been achieved (Scheme I) by refluxing N-cyclohexylhydroxylamine³ and 2, 3-dihydro-4*H*-pyran in dry benzene which were then trapped *in situ* by different dipolarophiles in a 1, 3-dipolar cycloaddition reaction with high regio- and stereoselectivity giving **2** (cf. Table I). Dimerization of nitrone could also be controlled under this condition⁴.

The concerted nature of these cycloaddition reactions with nitrone as 1, 3-dipole was generally accepted. The regioselectivity in these reactions was rationalized by using the frontier-orbital theory⁵. The ethyl acrylate adduct corresponded to this theory. There-

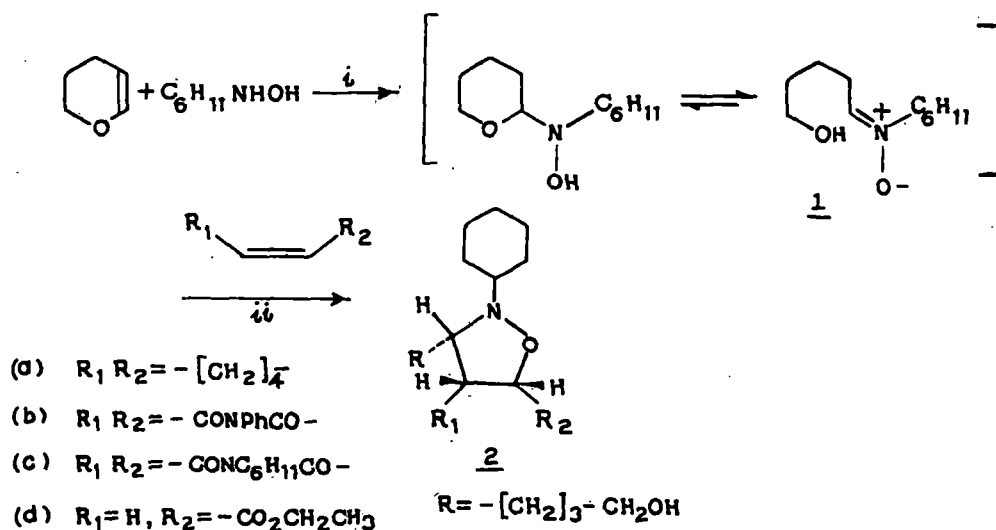
fore, the 5-substituted adduct for ethyl acrylate is due to LUMO (nitrone)-HOMO (dipolarophile) interaction.

Cycloaddition of *Z*-nitrone via an *exo*-transition state results in the formation of *syn*-isoxazolidine⁶. The relative configurations of C-3, C-4 and C-5 in the adducts were in favour of the *exo*-transition state geometry. The proton at C-3 and C-4 were *syn* in **2b** and **2c** and their coupling constants ($J_{3,4} = 6$ Hz) were also indicative of this stereochemical relationship, whilst a D₂O shake revealed the presence of one rapidly exchangeable proton at C-5. The stereochemical assignments of **2a** and **2d** at C-3, C-4 and C-5 on the isoxazolidine ring were also determined from ¹H NMR spectra. The coupling constant values ($J_{3,6} = 7.5-8.0$ Hz) revealed the formation of *syn*-isoxazolidine in both the cases via *exo*-transition state.

Further study of the reaction in various systems is in progress.

Table I—Physical data of the cycloadducts (**2**)

Cycloadduct	Yield (%)	R _f	m.p. °C
2a	33.63	0.30	106
2b	26.01	0.34	132
2c	29.50	0.38	114
2d	55.94	0.41	Dark red gummy liquid



Scheme I

Experimental

General procedure for cycloaddition

N-Cyclohexylhydroxylamine (2.17 mmoles) was added to a solution of 2, 3-dihydro-4H-pyran (1 equivalent) in dry benzene (20 ml) under nitrogen atmosphere and the reaction mixture refluxed for 24 hr. The reaction was monitored by TLC [silica gel; ethyl acetate-benzene (1:10)]. Dipolarophiles were added (1 equivalent) at this stage and the reaction mixture was further refluxed for 24 hr. The solvent was evaporated off and the cycloadducts [cf., Table 1] were isolated by column chromatography using benzene-pet. ether (60°-80°) as eluant.

Compound 2a: MS: m/z 281 (M^+), 113, 98; IR ($CHCl_3$): 3264 (br), 2924, 2851, 1540, 1443, 1248 cm^{-1} ; PMR ($CDCl_3$): δ 3.80-3.7 (d, 1H, $J=4.5$ Hz, C_5-H), 3.10-3.06 (br, 1H, $>CH-N<$), 2.63-2.46 (q, 1H, $J=7.5$ Hz, C_3-H), 2.36-2.16 (dd, 1H, $J=4.5$, $J=7.5$ Hz, C_4-H), 2.00-0.5 (m, 26H).

Compound 2b: MS: m/z 372 (M^+), 299, 289, 242, 173, 117, 113, 78; IR ($CHCl_3$): 3460 (br), 2940, 2840, 1780, 1700, 1690, 1480, 1395, 1190 cm^{-1} ; PMR ($CDCl_3$): δ 7.80-7.30 (m, 5H, C_6H_5), 5.70-5.45 (br, 1H, C_5-H , exchangeable with D_2O), 4.50-4.26 (dd, 1H, $J=6$ Hz, C_4-H), 3.38-3.10 (dd, 1H, $J=6$ Hz, C_3-H), 3.00-2.83 (m, 1H, $>CH-N<$), 2.16-0.43 (m, 18H).

Compound 2c: MS: m/z 378 (M^+), 323, 305, 295, 277, 267, 251, 208, 196, 170, 114, 83; IR ($CHCl_3$): 3580, 2920 (br), 1775, 1690, 1440, 1390, 1140, 890 cm^{-1} ; PMR ($CDCl_3$): δ 5.75-5.0 (br, 1H, $C(5H)H$, exchangeable,

with D_2O), 4.15-3.93 (dd, 1H, $J=6.06$ Hz, $J=6.06$, C_4-H), 3.20-3.10 (m, 1H, $>CH-N<$), 3.06-2.94 (dd, 1H, $J=6.06$ Hz, C_3-H), 2.93-2.84 (m, 1H, $>CH-N<O$), 2.20-1.03 (m, 28H).

Compound 2d: MS: m/z 299 (M^+), 296, 242, 226, 204, 187, 142, 131; IR ($CHCl_3$): 3340 (br), 2930, 2850, 1725, 1440 cm^{-1} ; PMR ($CDCl_3$): δ 4.77-4.64 (b, 1H, $C(5H)$, exchangeable with D_2O), 4.20 (q, 2H, $-OCH_2Me$), 3.10-2.98 (t, 1H, $J=8.1$ Hz, C_3-H), 2.98-2.80 (br, 1H, $>CH-N<$), 2.80-2.60 (q, 2H, $J=8.1$ Hz, C_4-H_2), 2.35-0.60 (m, $-CH_3$ and remaining protons).

Acknowledgement

The authors are thankful to the Principal, Sikkim Govt College, for providing laboratory facilities. Thanks are also due to CDRI, Lucknow, IIT, Bombay and RSIC, North Eastern Hill University, Shillong for providing spectral data. Financial support from UGC, New Delhi is also gratefully acknowledged.

References

- Breuer E, *The chemistry of amino nitroso and nitro compounds*, edited by S Patai (John Willy), 1982, 459.
- Murahashi S I, Mitsui H, Watanabe T & Zenki S, *Tetrahedron Lett*, 24 (1983) 1049; and references cited therein.
- Ghosh A R, *Indian J Chem*, 23B (1984) 449.
- Yu Y, Ohno M & Eguchi S, *Tetrahedron*, 49 (1993) 824.
- (a) Sustmann R, *Pure Appl Chem*, 40 (1974) 569.
(b) Houk K N, Sims J, Watts C R & Luskus L, *J Am Chem Soc*, 95 (1973) 7301.
- DeShong P, Dicken C M, Leginus J M & Whittle R R, *J Am Chem Soc*, 106 (1984) 5598.