

Figure Captions

Chapter 1

- Fig.1.1. (a) Schematic Diagram of a Typical Physical Vapour Deposition System.  
 (b) Source, (i) basket, (ii) helix.
- Fig.1.2. Schematic diagram of an Electron beam Evaporation System  
 (a) Work accelerated gun.  
 (b) Bent beam electron gun.  
 A - LT Magnetic Focussing, F - Anode, H - Copper hearth, I - pole pieces, J - Water coated tube, K - permanent U-Magnet, L - Evaporant material.
- Fig.1.3. Schematic diagram of an MBE System.
- Fig.1.4. Schematic diagram of an ARE System.  
 A-Filament, E-Filament cover, F-Water cooled discharge chamber, D-Gas inlet, B-Anode, H-Magnets, S-Substrate, J-Substrate heater, C-Evaporator.
- Fig.1.5. Schematic diagram of an Ion Plating System.
- Fig.1.6. Principal regions in a low-pressure dc glow discharge.
- Fig.1.7. Schematic arrangements for various Sputtering variants;  
 (a) Diode, (b) Triode and (c) Magnetron  
 A-ANODE, T-TARGET, V-CHAMBER, S-SUBSTRATE, P-PLASMA, G-SPUTTER GAS.
- Fig.1.8. A typical chemical vapour deposition system.
- Fig.1.9. A typical Triple-Zone furnace for GaAs deposition.
- Fig.1.10. Schematic diagram of a Spray pyrolysis setup.

## Chapter 2

- Fig.2.1. Experimental arrangement for thin film deposition by the dip technique.
- Fig.2.2. Variation of thickness of typical  $\text{Al}_2\text{O}_3$  films with the speed of withdrawal of the substrate.
- Fig.2.3. Variation of thickness of typical  $\text{Al}_2\text{O}_3$  films with the concentration of the starting solution.
- Fig.2.4. Diffraction pattern of  $\text{Al}_2\text{O}_3$  film  
 (a) on glass substrate.  
 (b) on mica substrate.
- Fig.2.5. An X-ray diffractogram of a typical  $\text{Al}_2\text{O}_3$  film deposited by the dip technique.
- Fig.2.6. An X-ray diffractogram of a typical  $\text{SnO}_2$  film deposited by the dip technique.
- Fig.2.7. A scanning Electron Micrograph of a typical  $\text{Al}_2\text{O}_3$  film deposited by the dip technique.

## Chapter 3

- Fig.3.1. Arrangements for the deposition of  $\text{SnO}_2$  films. Figure (c) shows the location of the  $\text{SnCl}_2$  band with respect to the substrate.
- Fig.3.2. An X-ray diffractogram of a typical  $\text{SnO}_2$  film.
- Fig.3.3. A scanning electron micrograph of a typical  $\text{SnO}_2$  film.
- Fig.3.4. Optical transmission versus wavelength for typical  $\text{SnO}_2$  films of various sheet resistance (a) Undoped: (---), 475  $\Omega/\square$  ; (-.-.-.-), 670  $\Omega/\square$  ; (—), 860  $\Omega/\square$  ; (.-.-.-), 1400  $\Omega/\square$  ; (.....) bare substrate. (b) Sb-doped (2 at.%): (-----), 110  $\Omega/\square$  ; (-.-.-.-), 550  $\Omega/\square$  ; (—) 870  $\Omega/\square$  ; (.....), bare substrate.

Fig.3.5. Sheet resistance — thickness relationship for undoped and 2 at.% Sb-doped  $\text{SnO}_2$  films prepared at various substrate temperatures :  $370^\circ\text{C}$ ,  $430^\circ\text{C}$  and  $400^\circ\text{C}$ .

Fig.3.6. Average (between 400 nm - 700 nm) optical transmission - sheet resistance relation for undoped and Sb-doped (2 at.%)  $\text{SnO}_2$  film on soda-glass substrate.

#### Chapter 4

Fig.4.1. Variation in thickness of CdS films on glass substrate with number of dippings ( $T_B = 400^\circ\text{C}$ ).

Fig.4.2. X-ray diffractograms of CdS thin films on glass substrate for different Cd:S ratios (double dipping;  $T_B = 400^\circ\text{C}$ ;  $\text{CuK}_\alpha$  radiation).

Fig.4.3. X-ray diffractograms of CdS, ZnS and  $\text{Cd}_{0.8}\text{Zn}_{0.2}\text{S}$  thin films on glass substrates (double dippings; M:S = 1:1;  $T_B = 400^\circ\text{C}$ ;  $\text{CuK}_\alpha$  radiation).

Fig.4.4. (a) Electron micrograph of CdS thin films on glass substrates with Cd:S ratios 1:2, 2:1, and 1:1, single dipping, deposited at baking temperature,  $T_B = 250^\circ\text{C}$ .

Fig.4.4. (b) Electron micrograph of CdS thin films on glass substrates with Cd:S ratios 1:2, 2:1 and 1:1, single dipping, deposited at baking temperature,  $T_B = 400^\circ\text{C}$ .

Fig.4.5. Electron micrograph of CdS thin films on glass substrates for various deposition conditions. (Cd:S = 1:1), single dipping, (a) baking temperature  $T_B = 200^\circ\text{C}$ , (b)  $T_B = 300^\circ\text{C}$ , (c)  $T_B = 400^\circ\text{C}$ .

Fig.4.6. Electron micrograph of CdS thin films on glass substrates for various deposition conditions (Cd:S = 1:1, baking temperature  $T_B = 400^\circ\text{C}$ ) (a) single dipping; (b) double dipping; (c) five dippings; (d) ten dippings.

Fig.4.7. Electron micrographs of CdS thin films on different substrates (Cd:S = 1:1;  $T_B = 400^\circ\text{C}$ ) : (a) on glass, double dipping; (b) on aluminium

sheet, double dipping; (c) on conducting tin oxide, single dipping; (d) on mica, double dipping.

Fig.4.8. (a) Electron micrograph of CdS films obtained with prebaking. Prebaking temperature = 70 °C, Final baking temperature = 400 °C.

Fig.4.8. (b) Electron micrograph for CdS film obtained by slow introduction of the substrate + liquid film into the furnace at a speed of .56 mm/sec.

Fig.4.9. Electron micrographs of (a) ZnS and (b)  $\text{Cd}_{0.8}\text{Zn}_{0.2}\text{S}$  thin films deposited on glass substrates (double dipping; M:S = 1:1;  $T_B = 400^\circ\text{C}$ ).

Fig.4.10. Optical absorption curve of CdS (ten dippings), ZnS (double dipping) and  $\text{Cd}_{0.8}\text{Zn}_{0.2}\text{S}$  (double dipping) films on soda-glass substrates (M:S = 1:1;  $T_B = 400^\circ\text{C}$ ).

## Chapter 5

Fig.5.1. Schematic diagram of Electron Multipliers (Discrete and Continuous dynode).

Fig.5.2. Schematic diagram of a typical Parallel-Plate Electron Multiplier (PPEM).

Fig.5.3. A typical vacuum system for study of Electron multiplier characteristics.

Fig.5.4. Mounting and electrical connections of Parallel-plate electron multiplier test circuit.

Fig.5.5. Parallel-plate electron multiplier preamplifier circuit.

Fig.5.6. A typical output pulse from the parallel-plate electron multiplier.

Fig.5.7. (a) A typical photoconductive (PC) cell geometry.  
(b) Test circuit for studying the photoresponse of a PC cell.

Fig.5.8. Scanning electron micrographs showing the effect of  $\text{CdCl}_2$  concentration in the sensitising

solution on the morphology of CdS thin films (baking time, 3 min) : (a)  $7.5 \times 10^{-2} \text{mol l}^{-1}$ , (b)  $12.5 \times 10^{-2} \text{mol l}^{-1}$ , (c)  $17.5 \times 10^{-2} \text{mol l}^{-1}$ .

- Fig.5.9. Voltage ——— Current (V-I) characteristics of a typical photoconductive cell for different illumination levels : line A, 1000 lx; line B, 200 lx; line C, 50 lx.
- Fig.5.10. Equivalent luminance ——— resistance ( $\bar{\Phi}$ -R) relation for a typical photoconductive cell, for two different baking times : Curve A, 7 min; Curve B, 3 min.
- Fig.5.11. Normalised spectral response of photoconductivity for pure CdS (unsensitised) film.
- Fig.5.12. Spectral response of photoconductivity for sensitised CdS films (normalised with respect to the maximum of the more sensitive sample) for two baking times : Curve A, 3 min; Curve B, 7 min.