LIST OF PUBLICATIONS

1. **Densities, Viscosities, Sound Speeds, Refractive Indices, and Excess Properties of Binary Mixtures of Isoamyl Alcohol with Some Alkoxyethanols**

Springer

**Published in International Journal of Thermophysics* (2010) 31, 316.
(Copy of Reprint Enclosed)

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2. **Study of the Solution Properties of Ternary Mixtures of 1, 3-Dioxolane (1), Diethyl Ether (2), and *n*-Amyl Alcohol (3) and the Corresponding Binary Mixtures by Density, Viscosity, Refractivity, and Ultrasonic Speed**

ACS Publications
High quality. High impact.

**Published in Journal of Chemical & Engineering Data* (2010) 55, 4536.
(Copy of Reprint Enclosed)

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3. **Ion Association and Solvation Behavior of Some Alkali Metal Acetates in Aqueous 2-butanol Solutions at *T* = 298.15, 303.15 and 308.15 K**

ELSEVIER

* *Published in Fluid Phase Equilibria* (2011) 307, 216.
(Galley Proof Enclosed)

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4. **Solute-Solvent and Solvent-Solvent Interactions of Menthol in Isopropyl Alcohol and its Binary Mixtures with Methyl Salicylate by Volumetric, Viscometric, Interferometric and Refractive Index Techniques.**

ELSEVIER

**Published in Journal of Thermochimica Acta* (2010) 499,149.
(Copy of Reprint Enclosed)

DEPARTMENT OF CHEMISTRY
University of Hengai

5. Ion- Pair and Triple- Ion Formation by Some Tetraalkylammonium Iodides in Binary Mixtures of Carbon tetrachloride + Nitro benzene

*Published in Journal of Chemical & Engineering Data, (2009) 54, 2429.
(Copy of Reprint Enclosed)

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6. Volumetric, Viscometric, Interferrometric and Refractometric Properties of 2-Methoxy Ethanol + Diethyl Ether +Dichloromethane Ternary System and its Corresponding Binaries at 298.15 K.

*Published in Physics and Chemistry of Liquids (2011) 49,133.
(Copy of Reprint Enclosed)

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7. Ion-Solvent and Ion-Ion Interactions of Sodium molybdate and Sodium tungstate in mixture of Ethane–1, 2 diol and Water at 298.15, 308.15 and 318.15K

*Published in Journal of Molecular Liquids (2009) 144, 149.
(Copy of Reprint Enclosed)

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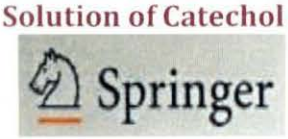
8. Ion-Solvent and Ion-Ion Interaction of Phosphomolybdic Acid in Aqueous Solution of Catechol at 298.15, 308.15, 318.15K



*Published in Russian Journal of Physical Chemistry A, (2009) 83, 18887.
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